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(54) Solid feeding of silane crosslinking agents into extruder

Zuführen von Festsilanvernetzungsmittel in einem Extruder

Alimentation en état solide d'un silane comme agent de réticulation dans une extrudeuse

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(56) References cited:
EP-A- 0 172 650 DE-A- 2 213 202
DE-A- 2 856 381

- CHEMICAL ABSTRACTS, vol. 96, no. 14, April
1982, Columbus, Ohio, US; abstract no. 105184Y,
page 32; && JP- A-56 155 213
- CHEMICAL ABSTRACTS, vol. 100, no. 10, March
1984, Columbus, Ohio, US; abstract no. 69488R,
page 52; && JP-A- 58 117 245

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Description**BACKGROUND OF THE INVENTION****5 Field of the Invention**

This invention relates to a process for crosslinking polymers (including homopolymers and copolymers). More particularly, the present invention relates to a process for silane crosslinking wherein the silane is fed into an extruder in a solid form.

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Description of the Prior Art

The properties of polymers can be modified by crosslinking of the polymer chains. One method for crosslinking which is commonly used in the industry is known as "silane crosslinking" i.e., crosslinking using olefinically unsaturated 15 alkoxy silanes. This method involves at least two stages. In the first stage, a silane is grafted onto base polymer chains in the presence of a free radical generator (e.g. peroxide catalyst) and in a second stage, crosslinking of the polymer chains occurs through the formation of Si-O-Si bonds as bridges. To effect crosslinking, the intermediate formed by grafting is subjected to moisture, for example, hot water or steam, in the presence of a hydrolysis/ condensation catalyst. Two processes most often employed are (1) a two-step Sioplas™ process as taught in U.S. Patent No. 3,646,155 20 and (2) a Monosil™ one-step process as taught in U.S. Patent No. 4,117,195.

A problem associated with both the Sioplas™ process and the Monosil™ process methods is the difficulty of adequately combining some or all of the silane, peroxide catalyst and hydrolysis/condensation catalyst (some or all of which are liquids) with the base polymer which is a solid. If the silane is injected as a liquid stream into a conventional extruder and mixed with a polymer, small gels form throughout the polymer product, apparently because of localized, premature 25 crosslinking in areas of high additive (silane and catalyst) concentration resulting from inadequate mixing.

When a silane is sprayed into a polymer, additional apparatus is required, and the silane must be sprayed uniformly. High demands are placed on the uniformity of mixing at or near the spraying point to equalize variations in concentration and ensure that the silane is distributed homogeneously.

To prevent localized, premature crosslinking in areas of high additive concentration and to equalize variations in 30 concentration at or near the point where the silane and peroxide are introduced, a long extruder, e.g. such as the one taught in U.S. Patent No. 4,117,195 may be employed. Furthermore, introducing the silane as a liquid into the conveying section of an extruder typically results in slippage and poor metering. The use of a long extruder also improves metering in such cases.

U.S. Patent No. 4,247,498 describes microporous products and a method for making microporous products which 35 can absorb relatively large amounts of "functionally useful liquids" and still behave as solids. The microporous products are made from synthetic thermoplastic polymers and "compatible liquids." In the course of producing the products, some of the compatible liquids are removed. The compatible liquids may also be functionally useful liquids and, when they are, the liquid remaining in the product permits the product to be used directly. A variety of such dual purpose liquids are disclosed at Column 27, line 62 to Column 28, line 32. Table VIII in this patent illustrates the formation of homogeneous porous polymer intermediates from polypropylene and compatible liquids. Footnote (1) suggests that 40 phenylmethyl polysiloxane was used as a compatible liquid. However, this patent does not mention using silane crosslinkers.

U.S. Patent No. 4,783,511 describes a process for the preparation of molded bodies from dry thermoplastic polyamides which have improved mechanical properties and higher shape-stability than the polyamide starting materials. 45 The process involves adding to the dry polyamide a masterbatch containing a silane and a "substrate" (i.e., a polyolefin, a swellable material or a non-swellable thermoplast sponge) before processing the dry polyamide. Three methods of masterbatch preparation described in the patent (Column 3, lines 11 ff.) are summarized as follows:

- 50 (1) A melt of a polyolefin is mixed with the silane and the silane-containing melt is drawn off as a cord which is comminuted to give a granulate which is carefully dried.
- (2) A silane-swellable particulate carrier material (for example, in the form of crumbs, grains or a coarse powder) is contacted with the silane, so that the silane is directly absorbed on the carrier. The carrier must remain sprayable, i.e., there should be no adhesion of the powder particulate.
- 55 (3) A thermoplast, which is in the form of a sponge with open pores but which has no swellability with respect to the silane, is mixed with the maximum amount of silane sufficient to fill the pores. The sprayability of the thermoplast should be essentially unchanged.

EP-A-0 172 650 discloses polyolefin blends comprising a mixture of a first polyethylene and minor amounts of a second solid polyethylene having incorporated therein an agent capable of reacting with polyolefin in a molten state.

Said reactive agent may comprise a cross-linking agent, e.g. a vinylalkoxysilane. The blends may be used for preparing blow-molded or extruded articles where exposure to water is not critical. Thus, the carrier of the reactive agent is not substantially free of water, and in some cases contact with water is even necessary to produce good adhesion to the substrates.

5 DE-A-28 56 381 relates to a process for modifying a filled polymer by grafting it with a silane compound which can be crosslinked by moisture. In this process, where a thermoplastic polyethylene, a solid carrier polymer, a vinylalkoxysilane and an organic peroxide are blended, the silane is soaked into the bulk of the polymer. Accordingly, extrusion equipment having high mixing capacity is required in order to ensure an even distribution of the silane throughout the extrudate.

10 **OBJECTIVES**

An object of the present invention is to provide a means for the convenient introduction of the silane crosslinker into a polymer.

15 Another object of the present invention is to avoid slipping problems during extrusion of crosslinkable polymer. Still another object of the present invention is to allow a silane crosslinker to be handled as a dry solid, i.e., capable of being dry fed into a mixing or compounding apparatus. A further object of the present invention is to allow for use in a standard-length extruder employed in the industry for grafting a silane onto a polymer.

20 An additional object of the present invention is to improve the physical properties of a fabricated product. Yet another object of the present invention is to reduce start-up time of the extrusion of the crosslinkable polymer by simplifying calibration of the silane feeding system.

Other objects of the invention will become apparent from the description and examples which follow.

25 **SUMMARY OF THE INVENTION**

The present invention provides a process as defined in claims 1 and 22 for crosslinking a base polymer with a silane where the silane has been introduced into (e.g., absorbed into or encapsulated by) a solid carrier polymer that is compatible with the base polymer. After the silane is introduced into the solid carrier polymer, the solid carrier polymer containing the silane is fed into the mixing apparatus where it is mixed with the base polymer. Subsequently, the mixture is exposed to moisture to effect crosslinking.

In a preferred embodiment, the silane is absorbed into a solid carrier polymer along with a free-radical generator, such as a peroxide, and/or a hydrolysis/condensation catalyst, usually a tin catalyst, are introduced into the mixing apparatus where it contacts the base polymer. In a most preferred embodiment other additives, such as stabilizers, and metal deactivators are also introduced into the solid carrier polymer along with the silanes and the catalysts and subsequently mixed with the base polymer in the mixing device (e.g., an extruder).

DETAILED DESCRIPTION OF THE INVENTION

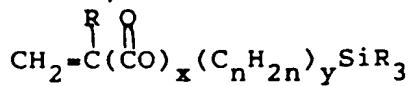
40 **Base Polymer**

The base polymer is the polymer to be modified by the process of the present invention. Such a base polymer is a thermoplastic polymer. The base polymer employed in the invention may be either a homopolymer of an alpha-olefin having 2 to 6 carbon atoms or a copolymer of two alpha-olefins, e.g., copolymers of ethylene; propylene; 1-butene; 1-pentene; 1-hexene; 1-octene; isobutylene; 2-methyl-1-butene; 3-methyl-1-butene; 2-methyl-1-pentene; 3-methyl-1-pentene; 4-methyl-1-pentene; 2,3-dimethyl-1-butene; 3,3-dimethyl-1-butene; and 2-ethyl-1-butene. Other polymers useful in this invention include ethylene vinyl acetate copolymer (EVA), ethylene alkyl acrylate copolymers such as ethylene ethylacrylate (EEA), and ethylene acrylic acid copolymer. Modified poly-alpha-olefins such as chlorinated polyethylene may also be useful in this invention. Mixtures of two or more base polymers may also be used.

50 The base polymer is added to the extruder as a solid, generally in pellet or granular form. Preferably the base polymer is polyethylene or a copolymer of ethylene and one of the above monomers. Most preferably, the base polymer is homopolymer of ethylene or a copolymer of ethylene with 1-butene, 1-hexene, 1-octene, or 4-methyl-1-pentene.

Silane

55 Silanes suitable for grafting onto and cross linking with a base polymer according to the present invention include silanes of the general formula:



wherein: R' represents a hydrogen atom or methyl group; x and y are 0 or 1 with the proviso that when x is 1, y equals 1; n is an integer from 1 to 12 inclusive, preferably 1 to 4; and each R independently represents a hydrolyzable organic group such as an alkoxy group having 1 to 12 carbon atoms (e.g. methoxy, ethoxy, butoxy), aryloxy group (e.g. phenoxy), aralkoxy (e.g. benzylxy), aliphatic acyloxy group having 1 to 12 carbon atoms (e.g. formyloxy, acetylxy, propenoxyloxy), oximo or substituted amino groups (alkylamino and arylamino), or a lower alkyl group having 1 to 6 carbon atoms inclusive, with the proviso that not more than one of the three R groups is an alkyl.

Some representative examples of the unsaturated silanes useful in the present process are vinyltrithoxysilane, 15 vinyltrimethoxysilane, allyltriethoxysilane, vinyl-tri-n-butoxysilane, hexenyltri-iso-butoxysilane, allyltri-n-pentoxysilane, dodecenyli-tri-n-octoxysilane, heptenyltri-n-heptoxy silane, allyltri-iso-propoxysilane, pentenyltri-n-propoxysilane, sec-butenyltriethoxysilane, and 3-methacryloxypropyl-trimethoxysilane.

Preferably the silanes used in the process of the present invention will contain a vinyl or methacryloxypropyl group as the group reacting with the free radical sites formed in the polymer by the free radical generator. These silanes are 20 commercially available from Union Carbide Chemical and Plastics Company Inc., e.g. A-171, A-151, A-174. The most preferred silane is vinyl trimethoxysilane.

The amount of silane is chosen so as to be near the minimum amount needed to provide the desired degree of crosslinking. The proportion of hydrolyzable silane based on the weight of the base polymer, e.g., polyethylene, ranges 25 from 0.1 to 10 wt. %, preferably 0.7 to 3 wt. %, of silane based on the total weight of polymer.

Carrier Polymer

Carrier polymers suitable for use in the present invention must be solids that are in particulate form and that are compatible with the polymer being crosslinked and the silane. By compatible is meant that the carrier polymer must not 30 readily react with the silane and must be dispersible or soluble in the base polymer. Preferably suitable carrier polymers are non-hygroscopic, that is, absorption of moisture is comparatively low to minimize the possibility of premature hydrolysis and condensation of the silane. In any event, the carrier polymer should be substantially free of water. In general, carrier polymers of the present invention are particulates in the form of powder, granules, or pellets. A preferred form is pellet.

35 The carrier polymer must be capable of incorporating the silane crosslinker while still retaining its particulate and solid characteristics. The three classes of carrier polymers are porous, sponge-like carrier polymers, swellable polymers and encapsulates.

Porous polymers are capable of incorporating the silane in the pores. Porous sponge-like carrier polymers suitable 40 for taking up silane may be prepared, for example, from various high and low density polyethylenes and polypropylenes. Preferred porous polymers for use in the present invention as carrier polymers include ethylene vinyl acetate copolymer (EVA), high density polyethylene, low density polyethylene and linear low density polyethylene. The pore size is small enough and the pore volume is large enough to hold a relatively large volume of silane. The pore volume is generally from 10-90% preferably about 30-90%; the cross-section of the pores is generally in the range of 0.1-5 μm and the cell size is generally from 1-30 μm . These porous polymers can take up 0.5 to three times their weight of silane. The porous 45 polymers may be employed as carrier polymers in powder, granular, or pellet form. However, the preferred form is pellet. Suitable porous polymers are commercially available and can be obtained from ENKA AG, Accurel Systems, Postfach, 8753 Obernberg, FRG, or prepared as taught in U.S. Patent No. 4,247,498.

50 Swellable polymers are capable of incorporating the silane by being swelled by the silane. The carrier polymer may also be chosen from polymers that are readily swelled by the silane and optionally the peroxide, hydrolysis/condensation catalyst, stabilizers, and other additives where these can be mixed with or dissolved in the silane to form a liquid mixture. One polymer suitable for this purpose is EVA, especially EVA having a high vinyl acetate content ranging from about 18 to 45 percent by weight. Such a swellable carrier polymer may be used in granule, powder, pellet, or other solid form.

The carrier polymer should be chosen so that the amount of silane that it can absorb without becoming wet or sticky 55 is a minimum of about 10% by weight. In practice it has been found that suitable swelled pellets containing about 20% vinyltrimethoxysilane can be prepared from EVA made from 26% vinyl acetate monomer. Polyethylene is not generally suitable as a swellable carrier polymer because it does not readily absorb sufficiently large amounts of silanes.

A third class of carrier polymer is an encapsulate. The silane is encapsulated, i.e., contained with a thermoplastic polymer capsule. Suitable polymers useful as encapsulates in the present invention are polyolefins. Suitable polyolefins

may be either a homopolymer of an alpha-olefin having 2 to 6 carbon atoms or a copolymer of two alpha-olefins. For example, encapsulation of the silane in the carrier polymer would produce a suitable solid form of the silane.

5 The amount of carrier polymer is selected to be the minimum amount needed to contain the desired amount of silane and optionally other additives in a dry, easily handled form. The carrier polymer is employed in amounts ranging from 0.1 to 20 weight percent, preferably, in amounts ranging from 0.7 to 7 weight percent, based on the weight of the extrudate.

Free-Radical Generator

10 The free radical generator used in the process of the present invention may be any compound that is capable of producing free radical sites in the base polymer, the reaction conditions depending on temperature and retention time needed for achieving a suitable half life time. Suitable free-radical initiators are organic peroxides and peresters such as *tert*-butylperoxypivalate, *tert*-butylperoxy-2-ethylhexanoate, dicumylperoxide, α , α' -bis (*tert*-butylperoxydiisopropyl)benzene, di-*tert*-butylperoxide, *tert*-butyl cumyl peroxide, di-benzoylperoxide, 2,5-dimethyl-2,5-bis (*tert*-butylperoxy)hexane, *tert*-butyl perbenzoate, and combinations thereof. Also an azo compound such as 2,2-azo-bis(2,4-dimethylvaleronitrile) and azo-bis-iso-butynitrile may be used as a free radical generator in the present invention.

15 Most preferred free radical generators are the peroxides, especially dicumylperoxide and α,α -bis(*tert*-butylperoxy)di-isopropylbenzene.

20 The amount of free radical generator necessary in the present invention is not narrowly critical and can be varied over wide ranges, for example, from 0.01 wt. % to 0.4 wt. %, preferably 0.05 to 0.2 wt. %, based on the total weight of the extrudate, depending on the nature of the base polymer, the silane, the presence of stabilizers, the extrusion conditions and so forth.

25 The amount of free radical generator needed depends on the identity and purity of the specific peroxide. For example, if it were found that a 1.5% loading of pure dicumyl peroxide (5.87% active oxygen) were sufficient in a formulation, but it were desired to try 2,5-dimethyl-2,5-di (*tert*-butylperoxy)hexane, available as a 91% active liquid (10% active oxygen), then one might initially estimate that the proper amount of the latter peroxide would be only $1.5\% \times (5.87/10)$, or 0.88% in the formulation.

Hydrolysis/Condensation Catalyst

30 Hydrolysis/condensation catalysts used in the process of the present invention catalyze the crosslinking of the extrudate by reaction of the extrudate with water. The catalysts may accelerate either the hydrolysis reaction of the grafted silyl groups with water to form silanols, or the condensation of silanols to form Si-O-Si bonds, or both. These catalysts may be Lewis acids, such as metal carboxylates, for example: dibutyltin dilaurate, dioctyltin dilaurate, stannous acetate, stannous octoate, dibutyltin dioctoate, di-octyl tin-bis (isooctylmaleate), di-octyl-tin-bis (isooctylthioglycolate), as well as organometal compounds such as titanium esters and chelates, for example, tetrabutyl titanate, tetranonyl titanate, and bis (acetylacetonyl) di-isopropyl titanate; organic bases, such as, triethylamine, hexylamine, dibutylamine, and piperidine; and protic acids, such as, fatty acids and mineral acids. Preferred hydrolysis/condensation catalysts include dibutyltin dilaurate (DBTDL), dioctyltin dilaurate (DOTDL) and stannous octoate.

40 The proportion of silanol condensation catalyst used is not narrowly critical, illustratively ranging from 0.01 to 0.2 wt. %, preferably 0.02 to 0.1 wt. %, based on the total weight of the extrudate.

45 When it is desired to store the crosslinkable polymer for some time prior to fabrication and crosslinking, then a hydrolysis/condensation catalyst should normally not be added during the production of the crosslinkable polymer. Instead, the hydrolysis/condensation catalyst should be mixed with the crosslinkable polymer at the fabrication step. However, if it is desired to perform the production and fabrication of the crosslinkable polymer in a single step, then the hydrolysis/condensation catalyst should be included. An extruder would generally be used as the mixing compounding apparatus in this case.

Additives

50 In addition to the silane, catalyst and free radical generator, additives which are commonly utilized in crosslinking may also be employed in the process of the present invention. These include any of the antioxidants and heat stabilizers commonly used in the processing of base polymers and combinations thereof. In addition, minerals for improving flame retardancy or as an internal source of water for the crosslinking, e.g. alumina trihydrate, zeolites or mineral fillers like chalk, talc, mica, silica, silicates or carbon black may be used.

55 Generally, absorption of the silane, alone or with other additives in liquid form, into the carrier polymer in the process of the present invention is accomplished by tumble blending, the carrier polymer, silane, and, optionally, the other additives together. Tumble blending, for example, may be accomplished in a Conus blender. If not all additives are liquid, then any solid components should first be dissolved in the silane. The blending is carried out under a blanket of nitro-

gen, carbon dioxide, or dry air in a closed system to maintain the silane substantially free of water and to minimize evaporation of the liquid ingredients. Optionally, during blending heat may be applied. The container in which the blending takes place must be non-reactive with the silane and other additives. The absorption of silane and any other liquid additive into the carrier polymer is carried out prior to feeding the silane into the mixing or compounding apparatus. Additives absorbed into the carrier polymer along with the silane may be incorporated, for example, at 1 to 50 percent of the weight of the carrier polymer.

Temperature

10 In accordance with the process of this invention, the process temperature generally ranges above the crystalline melting point of the base polymer, i.e., between 120°C and the degradation temperature of the polymer. Preferably, the temperature of the process ranges from 150°C to 225°C. The actual processing temperature employed will normally be determined by consideration of polymers being processed and the type of apparatus in which the process is performed.

15 Apparatus

20 The process of the present invention may be carried out employing any suitable apparatus. Preferably, however, the process is carried out under conditions in which the base polymer and solid carrier polymer containing silane are subjected to mechanical working, such as kneading or compounding. The process is, therefore, carried out in, for example, an extruder. Common extruders are the single or twin screw type. Other apparatus that can be employed can include a Buss Cokneader or a Banbury mixer. Such compounding equipment may be preferred to an extruder where the grafting reaction is to be performed and then the crosslinked polymer is to be stored for a period of time prior to fabrication.

25 All parts and percentages shown in the following examples are by weight unless otherwise specified.

25 DEFINITIONS

30 LLDPE linear low density polyethylene
LDPE low density polyethylene
PE polyethylene
HDPE high density polyethylene
EVA ethylene/vinyl acetate copolymer

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LIST OF MATERIALS USED

<u>Material Name</u>	<u>Description</u>
5 Silane 1	Vinyltriethoxysilane, available from Union Carbide Chemicals and Plastics Company Inc. as A-151.
10 Silane 2	Vinyltrimethoxysilane, available from Union Carbide Chemicals and Plastics Company Inc. as A-171.
15 BHT	2,6- <u>tert</u> -butyl-4-methylphenol, a stabilizer
20 Compound A	8% DICUP 88.95% Silane 1 3.0% DBTDL 0.5% BHT
25 Compound B	polyoxyethylene-grafted polydimethylsiloxane, a processing aid
30 Compound C	40% Porous Polymer B 48.6% Silane 2 2.4% Compound B 2.4% Trigono _x 101 0.49% DOTDL 4.3% Irganox 1010 1.8% Hostanox VPO SP1
35 Compound D	74.87% Silane 2 5.33% DICUP 1.84% DBTDL 5.12% Compound B 5.15% Hostanox VPO SP1 7.69% Irganox 1076
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5	Compound E	75.5% Silane 2 5.2% DICUP 1.8% DBTDL 5.0% Compound B 12.5% Naugard 445
10	Compound F	90.9% Silane 2 3.64% Trigono _x 101 4.55% Compound B 0.91% DOTDL
15	Compound G	80.97% Silane 2 4.04% Compound B 4.04% Trigono _x 101 0.81% DOTDL 7.08% Irganox 1010 3.06% Hostanox VPO SP1
20	Compound H	90.0% LLDPE (octene comonomer, 0.939 g/cm ³ , MI 2.5) 2.5% Naugard XL-1 7.5% Ethano _x 330
25	Compound I	40% Porous Polymer A 45% Silane 2 3.2% DICUP 1.1% DBTDL 3.1% Compound B 3.1% Hostanox VPO SP1 4.6% Irganox 1076
30	Compound J	40% Porous Polymer A 45.3% Silane 2 3.12% DICUP 1.08% DBTDL 3.0% Compound B 7.5% Naugard 445
35	Compound K	40% Porous Polymer A 54.54% Silane 2 2.18% Trigono _x 101 2.73% Compound B 0.54% DOTDL
40	Compound L	1.7 parts DICUP and 18.35 parts Silane 2 absorbed in 80 parts EVA-1
45		
50		

	Compound M	1 part DBTDL absorbed in 99 parts PE-1
5	Compound N	60 parts Compound A absorbed in 40 parts Porous Polymer A
10	DBTDL	dibutyltin dilaurate, hydrolysis/ condensation catalyst
15	DICUP	dicumyl peroxide
20	DOTDL	dioctyltin dilaurate, hydrolysis condensation catalyst
25	EVA-1	EVA, 26% VA content, MI 2 known as Escorene 26.2 available from Exxon
30	Ethanol 330	1,2,,4-trimethyl-2,3,6-tris[3,5-di-tertiary-butyl -4-hydroxybenzyl] benzene, available from Ethyl Corp.
35	Hostanox VPOSPI	tris[2- <u>tert</u> -butyl-4-thio(2'-methyl-4'-hydroxy-5'- <u>tert</u> -butyl)phenyl-5-methyl]phenyl phosphite from Hoeschst; a stabilizer
40	Irganox 1010	tetra-bis-methylene 3-(3,5-di- <u>tert</u> -butyl-4-hydroxyphenyl)propionate methane from Ciba-Geigy; a stabilizer
45	Irganox 1076	octadecyl 3-(3,5-di- <u>tert</u> -butyl-4-hydroxyphenyl)propionate methane from Ciba-Geigy; a stabilizer
50	PE-1	low density polyethylene (MI 1.2g/10 min, 0.922 g/cm ³) known as L-411 available from Statoil
55	PE-2	low density polyethylene (0.922 g/cm ³ , MI 1.8 g/10 min) known as L-420 available from Statoil
	PE-3	a high density polyethylene (0.953 g/cm ³ , MI 6.5 g/10 min (190°C, 2.16 kg)) known as Lupolen 5031-L available from BASF

5	PE-4	a linear low density polyethylene (octene comonomer, 0.939 g/cm ³ , MI 6.5 g/10 min (190°C, 2.16 kg)) known as Dowlex 2037E available from Dow Chemical
10	PE-5	a linear low density polyethylene (octene comonomer, 0.909 g/cm ³ , MI 8 g/10 min) known as Dowlex 3360E available from Dow Chemical
15	Naugard 445	4,4'-bis(α,α-dimethyl-benzyl) diphenylamine from Uniroyal; a stabilizer
20	Naugard XL-1	2,2'-oxamidobis[ethyl-3-(3,5-di- <u>tert</u> -butyl-4-hydroxyphenyl)- propionate; a stabilizer
25	Trigonox 101	2,5-dimethyl-2,5-di(<u>tert</u> -butyl- peroxy)hexane from Akzo
30	Porous Polymer A	porous LDPE known as Accurel available from Akzo Enka AG (Enka Industrial Products, Inc., Accurel Systems Group)
35	Porous Polymer B	porous HDPE known as Accurel available from Akzo Enka AG (Enka Industrial Products, Inc., Accurel Systems Group)

Test Methods:

40 The following test methods were used in the Examples:

45	Tensile Strength	ASTM D638
	Elongation (at break)	ASTM D638
	Gel Content	ASTM D2765
50	Hot set	IEC 540 (82) Art. 14 after 15 min. at 200°C, 20 N/cm ² load
	Permanent Elongation	after recovery for 5 min. at 200°C from Hot Set test

The following examples illustrate the present invention.

55 Example 1. Preparation of Compounds C, I, J, K, N

The vinylsilane and the other ingredients except the carrier polymer were combined and stirred at room temperature in a closed, dry glass or stainless steel vessel until a homogeneous solution in the vinylsilane was obtained. The carrier polymer, in pellet form, was placed in a dry, sealable cylindrical drum, filling the drum to about three-quarters

of its capacity. The liquid silane mixture was added and the drum was sealed. The drum was rotated at ambient temperature on motorized rubber rollers for 25-30 minutes, after which the liquid was completely absorbed in the carrier polymer. The pellets so obtained were stored in a dry atmosphere in a sealed container prior to use.

5 Example 2. Preparation of Compound L

The same procedure as described in Example 1 was followed, except that EVA pellets were substituted for the carrier polymer pellets, and the blending time necessary for complete absorption of the liquid was 2-3 hours.

10 Example 3. Preparation of Compound M

Pellets of polyethylene (99 parts) and DBTDL (1 part) were placed in a dry, sealed cylindrical container. The drum was rotated on mechanically rotated rollers in an oven at 80°C for one hour, after which time the DBTDL was fully absorbed in the polyethylene pellets.

15 Example 4 (comparative)

LDPE pellets were placed in a sealed 4-liter cylindrical drum together with the amount of Compound A indicated in Table 1, column 1 and 2. The drum was rotated on mechanically driven rollers at room temperature for 1-2 minutes. At 20 the end of this time, Compound A had uniformly coated the surface of the polyethylene pellets, but the pellets still appeared wet. This procedure simulates uniform spraying of Compound A into the throat of an extruder in a larger scale operation.

The silane-coated polyethylene pellets were extruded using a 36 mm 24/l K/D Leistritz extruder at a melt temperature of 210°C, through a 50 mm x 1 mm slit die, to form a tape. The tape was immersed in hot water for 4 hours at 90°C. 25 Physical properties were obtained as shown in Table 1, Columns 1 and 2.

Example 5.

Pellets of Polyethylene PE-1 and Compound N were tumble-blended in the ratio shown in Table 1, Columns 5 and 30 6. The physical mixture was formed into tape and crosslinked using the same conditions as in Example 4. The physical properties shown in Table 1, Columns 5 and 6, were obtained.

The amounts of silane, polymer, peroxide and catalyst employed in Examples 4 and 5 were identical. The physical properties obtained with Compound N were superior and the extruder output was superior to those obtained with the liquid/solid blend of Column 1. The falloff of rate seen in Column 2 is commonplace when liquid silanes are used at high 35 levels with polyethylene pellets in short extruders and was eliminated by the use of the present invention.

Example 6 (comparative)

Mixtures of polyethylene and EVA pellets were coated with Compound A in the ratios shown in Table 1, Columns 3 40 and 4, using the method of Example 4. The mixtures were extruded into tape and crosslinked as in Example 4. The physical properties shown in Table 1, Columns 3 and 4, were obtained.

Example 7.

45 Pellets of polyethylene, silane and peroxide absorbed in EVA, and hydrolysis/condensation catalyst absorbed in polyethylene were tumble-blended in the ratios shown in Table 1, Columns 7 and 8. The mixture was extruded into tape and crosslinked as in Example 5. The physical properties shown in Table 1, Columns 7 and 8, were obtained.

The relative amounts of polyethylene, EVA, silane, peroxide, and hydrolysis/condensation catalyst employed in 50 Examples 6 and 7 were identical. The physical properties shown in Table 1 showed improvements when the solid form of the silane/peroxide mixture was used.

Example 8.

Sleeves were extruded using an extruder and die intended for extruding jacketing onto electrical cable, as shown 55 in Table 2. After crosslinking by immersion in 90-95°C water for four hours, the hot set value of 80% indicated that a satisfactory level of crosslinking had been achieved.

Example 9.

High density polyethylene pipe was extruded using the equipment described in Table 3. The gel content values (obtained after immersion in 90-95°C water) showed that a satisfactory level of crosslinking had taken place.

5

Example 10.

Power cable insulation was extruded using the equipment and conditions described in Table 4. The change in physical properties after exposure to water demonstrates that satisfactory crosslinking was obtained using the present invention.

10

Table 1: Evaluation of Compositions

15 EQUIPMENT

Leistritz Extruder (36mm diameter, 24/1 length/diameter)

Screw design: Type B (barrier screw)

Die: 50mm x 1 mm slit

20

Temperature Profile (°C): 170 (feed)/190/210/210/210 (die)

25

INGREDIENTS	Comparative Composition				Composition of the Present Invention			
	1	2	3	4	5	6	7	8
PE-1	98.80	98.60	94.14	93.17	98.00	97.70	90.58	89.01
Compound N	-	-	-	-	2.00	2.30	-	-
Compound A	1.20	2.40	1.20	1.40	-	-	-	-
Compound L	-	-	-	-	-	-	5.82	6.79
EVA-1	-	-	4.66	5.43	-	-	-	-
Compound M	-	-	-	-	-	-	3.60	4.20

30

PHYSICAL PROPERTIES*

35

Tensile Strength, (MPa)	14.8	17.3	16	15.7	17.2	16.5	17.7	17.3
Elongation, (%)	598	667	602	623	865	756	790	619
Hot-Set, (%)	75	60	100	65	70	55	100	50
Output Extruder (kg/hr)**	4.31	3.63	4.63	4.52	4.57	4.72	4.69	4.46

40

* 4 hours crosslinking in water at 90°C

** after 20 minutes of extrusion, at 20 rpm

45

50

55

Table 2: Evaluation of Low-Density Polyethylene Sleeves on a Cable Extruder for Crosslinking

5

EQUIPMENT

Troester Extruder (90mm diameter, 25/1 length/diameter)

Spiral grooved for 1.5 mm diameters

10

Screw design: TS-IA 9040/87 PVC

Head: Transverse with 2 screens of 0.057mm/0.025mm mesh size (250/600 mesh) and breaker plate

15

Die: 30.2mm

Nozzle: 27.5mm

Temperature Profile (°C) : 160/170/180/190/200/200/190/210/210

Cooling temperature of the feeding zone (°C) : 50

20

INGREDIENTS

PE-2 100%

25

Compound I 3.3%

Composition Number 9

30

RUN CONDITIONS

35 Screw Speed, rpm 20

Motor Power, kw 4.4

Output of Crosslinkable

Polyethylene, kg/hr 36.5

Melt Temperature, °C 209

40

TEST RESULTS

Hot Set

Elongation/permanent, (%) 80/0

45

Tensile Strength
(MPa)

16.4

Elongation, (%) 623

50

* after 4 hours crosslinking in water at 90-95°C

55

Table 3: Evaluation of Crosslinked High Density Polyethylene Pipes

5

EQUIPMENT

10

Troester Extruder (60mm diameter, 25/1 length diameter)

15

Spiral grooved for 1.5 diameters

20

Screw Design: TS-1A 90029

25

Head: Crosshead for pipe

30

Die: 22.5mm

35

Nozzle: 16.4mm

40

Temperature Profile (°C): 175 (feed)/185/190/210/210/210 (head)

45

Composition Number 10¹ 11² 12³

50

INGREDIENTS (wt.%)

55

PE-3 100.0

60

Compound C 3.7

65

PE-4 97.0 97.0

70

Compound H 3.0 3.0

75

Compound K 2.5 3.7

80

1. polyethylene sanitary pipe formulation.
2. polyethylene pipe formulation.
3. polyethylene cable insulation.

85

Table 3A

Evaluation of Crosslinked High Density Polyethylene Pipes (Continued)					
Composition No.	Screw Speed (rpm)	Output (kg/hr)	Motor Current A	Gel Content*	Remarks
10	25	19.0	12	68.0	Excellent surface
	50	35.5	16		
	75	51.0	16		
	30	-	11		
11	25	18.0	14	75.3	Excellent surface
	50	34.5	20		
	75	50.0	22.5		
12	28	-	15	76.9	Excellent surface
	25	16.0	11		
	50	32.5	15		
	75	41.5	17		
	30	-	12.5		

55

*after immersion in water at 95°C for 6 hr.

Table 4: Evaluation of Extrusion of Low Voltage Power Cable Insulation

EQUIPMENT

5 PVC Extruder (120mm diameter by 24/1 length/diameter) Rosendahl (Austria)

Screw design: PVC screw:BM Type

Compression ratio: 2/1

Extrusion head/die: tube on

10 Breaker plate plus screen: (3x40 mesh), (3 x 0.42 mm mesh size)

RESULTS

15	Composition No.	Compound J* (wt %)	Line Speed (m/min)	Temperature** Profile	Tensile Strength (Mpa)	Elongation at Break (%)	Hot Set Permanent (%)
	13 ^a	0	18	(1)	12.9	453	-
	14 ^a	2.4	18	(1)	12.1	435	-
	15 ^a	2.4	52	(2)	12.0	478	-
20	16 ^a	2.7	18	(2)	12.5	413	-
	17 ^a	3.0	18	(2)	12.6	428	-
	18 ^b	2.4	18	(1)	14.1	400	80/0
	19 ^b	2.4	18	(2)	12.8	263	40/0
25	20 ^b	2.7	52	(2)	15.7	385	50/0
	21 ^b	3.0	18	(2)	14.1	392	45/0

30 * dry blended with LDPE (MI=3) in cement mixer for five minutes; air-conveyed to extruder hoper

** (1) 120/140/160/180/180-180°C
(2) 120/140/170/190/200-200°C35 (a) measurements made directly after extrusion
(b) measurements made after crosslinking for 8 hours at 90°C in water

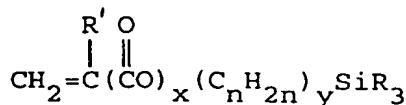
40

Claims

45 1. A process for grafting a silane onto a polymer in order to produce a polymer crosslinkable by exposure to moisture in the presence of a silane hydrolysis/condensation catalyst, said process comprising:

(a) blending

50 (i) a thermoplastic base polymer selected from (α) homopolymers of ethylene; (β) copolymers of ethylene and another α-olefin having 3 to 6 carbon atoms; (γ) ethylene-vinyl acetate copolymers, ethylene-ethylacrylate copolymers, ethylene-acrylic acid copolymers and mixtures thereof with each other or with copolymers of ethylene and 1-butene, 1-hexene, 1-octene or 4-methyl-1-pentene;
(ii) a solid carrier polymer selected from porous, sponge-like polymers, swellable polymers and encapsulates which do not readily react with the silane and are dispersible or soluble in the base polymer, said solid carrier polymer being substantially free of water and having contained therein a silane having the formula



10 wherein R' represents a hydrogen or a methyl group; x and y are 0 or 1, with the proviso that when x is 1, y equals 1; n is an integer from 1 to 12 inclusive; each R independently represents a hydrolyzable organic group selected from the group consisting of alkoxy, aryloxy, aralkoxy, aliphatic acyloxy, oximo, alkylamino and arylamino, or a lower alkyl group having from 1 to 6 carbon atoms inclusive, with the proviso that not more than one of the three R groups is an alkyl; wherein the amount of said carrier polymer ranges from 0.1 to 20 weight percent, based on the weight of the extrudate, and said silane being present in an amount ranging from 0.1 to 10 weight percent based on the total weight of the base polymer; and

15 (iii) a free radical generator;

(b) heating and mixing (i), (ii) and (iii) together at a temperature above the crystalline melting point of said base polymer.

20 2. A process according to claim 1 wherein said base polymer is a homopolymer of an alpha-olefin having 2 to 6 carbon atoms or a copolymer of two alpha-olefins, e.g. selected from the group consisting of copolymers of ethylene, propylene, 1-butene, 1-pentene, 1-hexene, 1-octene, isobutylene, 2-methyl-1-butene, 3-methyl-1-pentene, 4-methyl-1-pentene, 2,3-dimethyl-1-butene, 3,3-dimethyl-1-butene, 2-ethyl-1-butene, and mixtures thereof.

25 3. A process according to claim 1 wherein said base polymer is selected from the group consisting of ethylene vinyl acetate copolymer, ethylene ethyl acrylate copolymer, ethylene acrylic acid copolymer, and mixtures thereof with each other or with copolymers of ethylene and 1-butene, 1-hexene, 1-octene, or 4-methyl-1-pentene.

30 4. A process according to claim 1 wherein said base polymer is polyethylene or a copolymer of ethylene with 1-butene, 1-hexene, 1-octene, or 4-methyl-1-pentene.

5. A process according to at least one of the claims 1-4 wherein said carrier polymer is compatible with said base polymer and said silane.

35 6. A process according to at least one of the claims 1-5 wherein said carrier polymer is a porous polymer selected from the group consisting of high density polyethylene, low density polyethylene and linear low density polyethylene.

40 7. A process according to claim 6 wherein said porous polymer has a pore volume from 30 to 90 percent.

8. A process according to at least one of the claims 6-7 wherein said porous polymer is in pellet form.

9. A process according to at least one of the claims 1-8 wherein the amount of said carrier polymer ranges from 0.7 to 7 weight percent, based on the weight of the extrudate.

45 10. A process according to at least one of the claims 1-9 wherein said carrier polymer encapsulates said silane.

11. A process according to at least one of the claims 1-10 wherein said carrier polymer is readily swelled by said silane.

50 12. A process according to claim 11 wherein said carrier polymer is ethylene vinyl acetate copolymer.

13. A process according to at least one of the claims 1-12 wherein said silane is selected from the group consisting of vinyltriethoxysilane, vinyltrimethoxysilane, vinyltri-iso-butoxysilane.

55 14. A process according to at least one of the claims 1-13 wherein said silane is present in an amount ranging from 0.7 to 3 weight percent, based on the total weight of the base polymer.

15. A process according to at least one of the claims 1-14 wherein said free radical generator is selected from the

group consisting of dicumyl peroxide and α, α' -bis(tertiary-butylperoxy)di-isopropylbenzene.

16. A process according to claim 15 wherein said free radical generator is present in an amount ranging from 0.01 weight percent to 0.4 weight percent based on the total weight of the extrudate.

5 17. A process according to at least one of the claims 1-16 wherein said carrier polymer contains said silane and a free radical generator.

10 18. A process according to claim 17 wherein said carrier polymer contains other additives, e.g. a hydrolysis/condensation catalyst.

19. A process according to at least one of claims 1-18 wherein said hydrolysis/condensation catalysts are selected from the group consisting of dibutyltin dilaurate, diocetyltin dilaurate, and stannous octoate.

15 20. A process according to claim 19 wherein said hydrolysis/condensation catalyst is present in an amount ranging from 0.01 to 0.2 weight percent based on the total weight of the extrudate.

21. A process according to claim 1 comprising

20 (a) blending

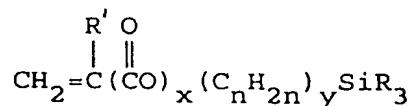
(i) a polyolefin polymer;
 (ii) a porous polymer compatible with said polyolefin and selected from the group consisting of high density polyethylene, low density polyethylene and linear low density polyethylene and wherein said porous polymer has a pore volume from 30 to 90 percent, a pore cross-section ranging from 0.1 to 5 μm , and a pore cell size ranging from 1 to 30 μm ; said porous polymer having contained therein a vinylalkoxysilane; and
 25 (iii) a free radical generator;

30 (b) heating and mixing (i), (ii) and (iii) together at a temperature above the crystalline melting point of said base polyolefin.

22. A process for producing crosslinked polymer comprising:

35 (a) blending

(i) a thermoplastic base polymer selected from (a) homopolymers of ethylene; (b) copolymers of ethylene and another α -olefin having 3 to 6 carbon atoms; (c) ethylene-vinyl acetate copolymers, ethylene-ethylacrylate copolymers, ethylene-acrylic acid copolymers and mixtures thereof with each other or with copolymers of ethylene and 1-butene, 1-hexene, 1-octene or 4-methyl-1-pentene;
 40 (ii) a solid carrier polymer selected from porous, sponge-like polymers, swellable polymers and encapsulates which do not readily react with the silane and are dispersible or soluble in the base polymer, said solid carrier polymer being substantially free of water and having contained therein a silane having the formula



45 50 wherein R' represents a hydrogen or a methyl group; x and y are 0 or 1, with the proviso that when x is 1, y equals 1; n is an integer from 1 to 12 inclusive; each R independently represents a hydrolyzable organic group selected from the group consisting of alkoxy, aryloxy, aralkoxy, aliphatic acyloxy, oximo, alkylamino and arylamino, or a lower alkyl group having from 1 to 6 carbon atoms inclusive, with the proviso that not more than one of the three R groups is an alkyl; wherein the amount of said carrier polymer ranges from 0.1 to 20 weight percent, based on the weight of the extrudate, and said silane being present in an amount ranging from 0.1 to 10 weight percent based on the total weight of the base polymer; and
 55 (iii) a free radical generator;

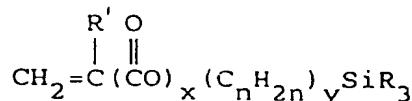
(b) heating and mixing (i), (ii) and (iii) together at a temperature above the crystalline melting point of said base polymer;
 (c) exposing the mixture of (b) to a hydrolysis/condensation catalyst and water.

5 Patentansprüche

1. Verfahren zum Pflöpfen eines Silans auf ein Polymer, um ein Polymer herzustellen, das bei Einwirkung von Feuchtigkeit in Anwesenheit eines Silan-Hydrolyse/Kondensations-Katalysators vernetzbar ist, wobei das Verfahren umfaßt:

10 (a) das Mischen

15 (i) eines thermoplastischen Grundpolymers, ausgewählt aus: (α) Homopolymeren von Ethylen; (β) Copolymeren von Ethylen und einem anderen α -Olefin mit 3 bis 6 Kohlenstoffatomen; (γ) Ethylen-Vinylacetat-Copolymeren, Ethylen-Ethylacrylat-Copolymeren, Ethylen-Acrysäure-Copolymeren und Mischungen von diesen miteinander oder mit Copolymeren von Ethylen und 1-Buten, 1-Hexen, 1-Octen oder 4-Methyl-1-penten;
 20 (ii) eines festen Trägerpolymers, ausgewählt aus porösen, schwammartigen Polymeren, quellbaren Polymeren und Einbettmaterialien, die sich mit dem Silan nicht leicht umsetzen und dispergierbar oder löslich in dem Grundpolymer sind, wobei das genannte feste Trägerpolymer im wesentlichen kein Wasser und ein Silan enthält mit der Formel



25 worin R' Wasserstoff oder eine Methylgruppe bedeutet, x und y = 0 oder 1 sind unter der Voraussetzung, daß y 1 ist wenn x 1 ist; n eine ganze Zahl einschließlich 1 bis 12 bedeutet; jede Gruppe R unabhängig voneinander eine hydrolysierbare organische Gruppe ist, ausgewählt aus Alkoxy, Aryloxy, Aralkyloxy, aliphatischem Acyloxy, Oximo, Alkylamino und Arylamino, oder eine niedere Alkylgruppe mit einschließlich 1 bis 6 Kohlenstoffatomen, unter der Voraussetzung, daß nicht mehr als eine der drei R-Gruppen ein Alkyl ist; wobei die Menge des genannten Trägerpolymers von 0,1 bis 20 Gew.-%, bezogen auf das Gewicht des Extrudats, beträgt und das genannte Silan in einer Menge von 0,1 bis 10 Gew.-%, bezogen auf das Gesamtgewicht des Grundpolymers, anwesend ist; und
 (iii) einer Erzeugerstanz für freie Radikale;

30 (b) das Erhitzen und Vermischen von (i), (ii) und (iii) bei einer Temperatur über der Kristallitschmelztemperatur des genannten Grundpolymers.

40 2. Verfahren nach Anspruch 1, wobei das genannte Grundpolymer ein Homopolymer eines α -Olefins mit 2 bis 6 Kohlenstoffatomen oder ein Copolymer von zwei α -Olefinen ist, z. B. ausgewählt aus der Gruppe: Copolymeren von Ethylen, Propylen, 1-Buten, 1-Penten- 1-Hexen, 1-Octen, Isobutylen, 2-Methyl-1-butlen, 3-Methyl-1-penten, 4-Methyl-1-penten, 2,3-Dimethyl-1-butlen, 3,3,-Dimethyl-1-butlen, 2-Ethyl-1-butlen und Mischungen von diesen.

45 3. Verfahren nach Anspruch 1, wobei das genannte Grundpolymer ausgewählt wird aus der Gruppe: Ethylen-Vinylacetat-Copolymer, Ethylen-Ethylacrylat-Copolymer, Ethylen-Acrysäure-Copolymer und Mischungen von diesen untereinander oder mit Copolymeren von Ethylen und 1-Buten, 1-Hexen, 1-Octen oder 4-Methyl-1-penten.

50 4. Verfahren nach Anspruch 1, wobei das genannte Grundpolymer ein Polyethylen oder ein Copolymer von Ethylen mit 1-Buten, 1-Hexen,- 1-Octen oder 4-Methyl-1-penten ist.

55 5. Verfahren nach wenigstens einem der Ansprüche 1-4, wobei das genannte Trägerpolymer mit dem genannten Grundpolymer und dem genannten Silan kompatibel ist.

6. Verfahren nach wenigstens einem der Ansprüche 1-5, wobei das genannte Trägerpolymer ein poröses Polymer ist, ausgewählt aus der Gruppe: Polyethylen mit hoher Dichte, Polyethylen mit geringer Dichte und linearem Polyethy-

len mit geringer Dichte.

7. Verfahren nach Anspruch 6, wobei das genannte poröse Polymer ein Porenvolumen von 30 bis 90 Prozent aufweist.

5 8. Verfahren nach wenigstens einem der Ansprüche 6-7, wobei das genannte poröse Polymer in pellet-Form vorliegt.

9. Verfahren nach wenigstens einem der Ansprüche 1-8, wobei die Menge des genannten Trägerpolymers von 0,7 bis 7 Gew.-%, bezogen auf das Gewicht des Extrudats, beträgt.

10 10. Verfahren nach wenigstens einem der Ansprüche 1-9, wobei das genannte Trägerpolymer das genannte Silan einkapselt.

15 11. Verfahren nach wenigstens einem der Ansprüche 1-10, wobei das genannte Trägerpolymer durch das genannte Silan leicht quellbar ist.

12. Verfahren nach Anspruch 11, wobei das genannte Trägerpolymer Ethylen-Vinylacetat-Copolymer ist.

20 13. Verfahren nach wenigstens einem der Ansprüche 1-12, wobei das genannte Silan ausgewählt wird aus der Gruppe: Vinyltriethoxysilan, Vinyltrimethoxysilan, Vinyltri-iso-butoxysilan.

14. Verfahren nach wenigstens einem der Ansprüche 1-13, wobei das genannte Silan in einer Menge von 0,7 bis 3 Gew.-%, bezogen auf das Gesamtgewicht des Grundpolymers, anwesend ist.

25 15. Verfahren nach wenigstens einem der Ansprüche 1-13, wobei die Erzeugersubstanz für freie Radikale ausgewählt wird aus der Gruppe: Dicumylperoxid und α,α' -Bis(tert.-butylperoxy)diisopropylbenzol.

16. Verfahren nach Anspruch 15, wobei die Erzeugersubstanz für freie Radikale in einer Menge von 0,01 bis 0,4 Gew.-%, bezogen auf das Gesamtgewicht des Extrudats, anwesend ist.

30 17. Verfahren nach wenigstens einem der Ansprüche 1-16, wobei das genannte Trägerpolymer das genannte Silan und eine Erzeugersubstanz für freie Radikale enthält.

35 18. Verfahren nach Anspruch 17, wobei das genannte Trägerpolymer andere Additive enthält, z. B. einen Hydrolyse/Kondensations-Katalysator.

19. Verfahren nach wenigstens einem der Ansprüche 1-18, wobei die genannten Hydrolyse/Kondensations-Katalysatoren ausgewählt werden aus der Gruppe: Dibutylzinn-dilaureat, Diocetylzinn-dilaureat und Zinn-II-octonoat.

40 20. Verfahren nach Anspruch 19, wobei der genannte Hydrolyse/Kondensations-Katalysator in einer Menge von 0,01 bis 0,2 Gew.-%, bezogen auf das Gesamtgewicht des Extrudats, anwesend ist.

21. Verfahren nach Anspruch 1, welches umfaßt:

45 (a) das Mischen

(i) eines Polyolefinpolymers;
 (ii) eines porösen, mit dem genannten Polyolefin kompatiblen Polymers, ausgewählt aus der Gruppe: Polyethylen mit hoher Dichte, Polyethylen mit geringer Dichte und linearem Polyethylen mit geringer Dichte, und wobei das genannte poröse Polymer ein Porenvolumen von 30 bis 90 Prozent, einen Porenquerschnitt von 0,1 bis 5 μm und eine Poren-Zellgröße von 1 bis 30 μm aufweist; wobei in dem genannten porösen Polymer ein Vinylalkoxysilan enthalten ist und
 (iii) einer Erzeugersubstanz für freie Radikale;

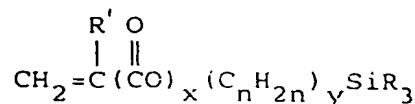
55 (b) das Erhitzen und Vermischen von (i), (ii) und (iii) bei einer Temperatur über der Kristallitschmelztemperatur des genannten Grundpolymers.

22. Verfahren zur Herstellung eines vernetzten Polymers, welches umfaßt:

(a) das Mischen

5 (i) eines thermoplastischen Grundpolymers, ausgewählt aus: (α) Homopolymeren von Ethylen; (β) Copolymeren von Ethylen und einem anderem α -Olefin mit 3 bis 6 Kohlenstoffatomen; (γ) Ethylen-Vinylacetat-Copolymeren, Ethylen-Ethylacrylat-Copolymeren, Ethylen-Acrylicsäure-Copolymeren und Mischungen von diesen miteinander oder mit Copolymeren von Ethylen und 1-Buten, 1-Hexen, 1-Octen oder 4-Methyl-1-penten;

10 (ii) eines festen Trägerpolymers, ausgewählt aus porösen, schwammartigen Polymeren, quellbaren Polymeren und Einbettmaterialien, die sich mit dem Silan nicht leicht umsetzen und dispergierbar oder löslich in dem Grundpolymer sind, wobei das genannte feste Trägerpolymer im wesentlichen kein Wasser und ein Silan enthält mit der Formel



20 worin R' Wasserstoff oder eine Methylgruppe bedeutet, x und y = 0 oder 1 sind unter der Voraussetzung, daß y 1 ist wenn x 1 ist; n eine ganze Zahl einschließlich 1 bis 12 bedeutet; jede Gruppe R unabhängig voneinander eine hydrolysierbare organische Gruppe ist, ausgewählt aus Alkoxy, Aryloxy, Aralkyloxy, aliphatischem Acyloxy, Oximo, Alkylamino und Arylamino, oder eine niedere Alkylgruppe mit einschließlich 1 bis 6 Kohlenstoffatomen, unter der Voraussetzung, daß nicht mehr als eine der drei R-Gruppen ein Alkyl ist; wobei die Menge des genannten Trägerpolymers von 0,1 bis 20 Gew.-%, bezogen auf das Gewicht des Extrudats, beträgt und das genannte Silan in einer Menge von 0,1 bis 10 Gew.-%, bezogen auf das Gesamtgewicht des Grundpolymers, anwesend ist; und

25 (iii) einer Erzeuger substanz für freie Radikale;

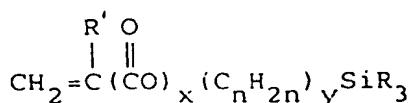
30 (b) das Erhitzen und Vermischen von (i), (ii) und (iii) bei einer Temperatur über der Kristallitschmelztemperatur des genannten Grundpolymers;

(c) die Einwirkung eines Hydrolyse/Kondensations-Katalysators und Wasser auf die Mischung von (b).

(a) das Mischen

35 (i) eines thermoplastischen Grundpolymers, ausgewählt aus: (α) Homopolymeren von Ethylen; (β) Copolymeren von Ethylen und einem anderem α -Olefin mit 3 bis 6 Kohlenstoffatomen; (γ) Ethylen-Vinylacetat-Copolymeren, Ethylen-Ethylacrylat-Copolymeren, Ethylen-Acrylicsäure-Copolymeren und Mischungen von diesen miteinander oder mit Copolymeren von Ethylen und 1-Buten, 1-Hexen, 1-Octen oder 4-Methyl-1-penten;

40 (ii) eines festen Trägerpolymers, ausgewählt aus porösen, schwammartigen Polymeren, quellbaren Polymeren und Einbettmaterialien, die sich mit dem Silan nicht leicht umsetzen und dispergierbar oder löslich in dem Grundpolymer sind, wobei das genannte feste Trägerpolymer im wesentlichen kein Wasser und ein Silan enthält mit der Formel



50 worin R' Wasserstoff oder eine Methylgruppe bedeutet, x und y = 0 oder 1 sind unter der Voraussetzung, daß y 1 ist wenn x 1 ist; n eine ganze Zahl einschließlich 1 bis 12 bedeutet; jede Gruppe R unabhängig voneinander eine hydrolysierbare organische Gruppe ist, ausgewählt aus Alkoxy, Aryloxy, Aralkyloxy, aliphatischem Acyloxy, Oximo, Alkylamino und Arylamino, oder eine niedere Alkylgruppe mit einschließlich 1 bis 6 Kohlenstoffatomen, unter der Voraussetzung, daß nicht mehr als eine der drei R-Gruppen ein Alkyl ist; wobei die Menge des genannten Trägerpolymers von 0,1 bis 20 Gew.-%, bezogen auf das Gewicht des Extrudats, beträgt und das genannte Silan in einer Menge von 0,1 bis 10 Gew.-%, bezogen auf das

Gesamtgewicht des Grundpolymers, anwesend ist; und
 (iii) einer Erzeuger substanz für freie Radikale;

5 (b) das Erhitzen und Vermischen von (i), (ii) und (iii) bei einer Temperatur über der Kristallitschmelztemperatur
 des genannten Grundpolymers;
 (c) die Einwirkung eines Hydrolyse/Kondensations-Katalysators und Wasser auf die Mischung von (b).

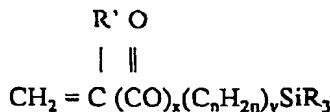
Revendications

10 1. Procédé de greffage d'un silane sur un polymère de manière à produire un polymère réticulable par exposition à l'humidité en présence d'un catalyseur d'hydrolyse/condensation de silane, ledit procédé comprenant les étapes suivantes :

15 (a) on mélange

20 (i) un polymère de base thermoplastique sélectionné parmi (α) les homopolymères d'éthylène; (β) les copolymères d'éthylène et d'une autre α -oléfine ayant 3 à 6 atomes de carbone; (γ) les copolymères d'éthylène et d'acétate de vinyle, les copolymères d'éthylène et d'acrylate d'éthyle, les copolymères d'éthylène et d'acide acrylique et leurs mélanges les uns avec les autres et avec des copolymères d'éthylène et de 1-butène, de 1-hexène, de 1-octène ou de 4-méthyl-1-pentène;
 (ii) un polymère véhiculaire solide sélectionné parmi les polymères poreux de type éponge, les polymères gonflables et les encapsulats qui ne réagissent pas aisément avec le silane et sont dispersables ou solubles dans le polymère de base, ledit polymère véhiculaire solide étant sensiblement exempt d'eau et contenant un silane ayant pour formule :

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35 dans laquelle R' représente de l'hydrogène ou un groupe méthyle; x et y sont égaux à 0 ou à 1, pourvu que, lorsque x est égal à 1, y soit égal à 1 également; n est un nombre entier de 1 à 12 inclus; chaque groupe R représente indépendamment un groupe organique hydrolysable sélectionné parmi les groupes alcoxy, aryloxy, aralcoxy, acyloxy aliphatique, oximo, alkylamino et arylamino ou un groupe alkyle inférieur ayant 1 à 6 atomes de carbone inclus, pourvu qu'au maximum un des trois groupes R soit un groupe alkyle; la quantité dudit polymère véhiculaire se siluant dans une plage de 0,1 à 20 pour-cent en poids sur la base du poids de l'extrudat et ledit silane étant présent en quantité de 0,1 à 10 pour-cent en poids sur la base du poids total du polymère de base; et
 40 (iii) un générateur de radicaux libres;

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(b) on chauffe et on mélange (1), (ii) et (iii) conjointement à une température supérieure au point de fusion cristalline dudit polymère de base.

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2. Procédé selon la revendication 1, dans lequel ledit polymère de base est un homopolymère d'une α -oléfine ayant 2 à 6 atomes de carbone ou un copolymère de deux α -oléfines, sélectionné par exemple parmi les copolymères d'éthylène, de propylène, de 1-butène, de 1-pentène, de 1-hexène, de 1-octène, d'isobutylène, de 2-méthyl-1-butène, de 3-méthyl-1-pentène, de 4-méthyl-1-pentène, de 2,3-diméthyl-1-butène, de 3,3-diméthyl-1-butène, de 2-éthyl-1-butène et de leurs mélanges.
3. Procédé selon la revendication 1, dans lequel ledit polymère de base est sélectionné parmi les copolymères d'éthylène et d'acétate de vinyle, les copolymères d'éthylène et d'acrylate d'éthyle, les copolymères d'éthylène et d'acide acrylique et leurs mélanges les uns avec les autres et avec des copolymères d'éthylène et de 1-butène, de 1-hexène, de 1-octène ou de 4-méthyl-1-pentène.
4. Procédé selon la revendication 1, dans lequel ledit polymère de base est le polyéthylène ou un copolymère d'éthylène avec le 1-butène, le 1-hexène, le 1-octène ou le 4-méthyl-1-pentène.

5. Procédé selon au moins l'une des revendications 1 à 4, dans lequel ledit polymère véhiculaire est compatible avec ledit polymère de base et ledit silane.

5 6. Procédé selon au moins l'une des revendications 1 à 5, dans lequel ledit polymère véhiculaire est un polymère poreux sélectionné parmi le polyéthylène de haute densité, le polyéthylène de basse densité et le polyéthylène linéaire de basse densité.

10 7. Procédé selon la revendication 6, dans lequel ledit polymère poreux a un volume de pores de 30 à 90 pour-cent.

10 8. Procédé selon au moins l'une des revendications 6 et 7, dans lequel ledit polymère poreux se présente sous la forme de pastilles.

15 9. Procédé selon au moins l'une des revendications 1 à 8, dans lequel la quantité dudit polymère véhiculaire se situe dans une plage de 0,7 à 7 pour-cent en poids sur la base du poids de l'extrudat.

15 10. Procédé selon au moins l'une des revendications 1 à 9, dans lequel ledit polymère véhiculaire encapsule ledit silane.

20 11. Procédé selon au moins l'une des revendications 1 à 10, dans lequel ledit polymère véhiculaire est aisément gonflé par ledit silane.

20 12. Procédé selon la revendication 11, dans lequel ledit polymère véhiculaire est un copolymère d'éthylène et d'acétate de vinyle.

25 13. Procédé selon au moins l'une des revendications 1 à 12, dans lequel ledit silane est sélectionné parmi le vinyltrithoxysilane, le vinyltriméthoxysilane et le vinyltri-iso-butoxysilane.

25 14. Procédé selon au moins l'une des revendications 1 à 13, dans lequel ledit silane est présent en quantité se situant dans la plage de 0,7 à 3 pour-cent en poids sur la base du poids total du polymère de base.

30 15. Procédé selon au moins l'une des revendications 1 à 14, dans lequel ledit générateur de radicaux libres est sélectionné parmi le peroxyde de dicumyle et l' α,α' -bis(*tert*-butylperoxy)di-isopropylbenzène.

30 16. Procédé selon la revendication 15, dans lequel ledit générateur de radicaux libres est présent en quantité de 0,01 à 0,4 pour-cent en poids sur la base du poids total de l'extrudat.

35 17. Procédé selon au moins l'une des revendications 1 à 16, dans lequel ledit polymère véhiculaire contient ledit silane et un générateur de radicaux libres.

40 18. Procédé selon la revendication 17, dans lequel ledit polymère véhiculaire contient d'autres additifs, par exemple un catalyseur d'hydrolyse/condensation.

40 19. Procédé selon au moins l'une des revendications 1 à 18, dans lequel lesdits catalyseurs d'hydrolyse/condensation sont sélectionnés parmi le dilaurate de dibutylétain, le dilaurate de dioctylétain et l'octonoate stanneux.

45 20. Procédé selon la revendication 19, dans lequel ledit catalyseur d'hydrolyse/condensation est présent en quantité de 0,01 à 0,2 pour-cent en poids sur la base du poids total de l'extrudat.

50 21. Procédé selon la revendication 1, comprenant les étapes suivantes:

50 (a) on mélange

55 (i) un polymère de polyoléfine;

55 (ii) un polymère poreux compatible avec ladite polyoléfine et sélectionné parmi le polyéthylène de haute densité, le polyéthylène de basse densité et le polyéthylène linéaire de basse densité, et dans lequel ledit polymère poreux a un volume de pores de 30 à 90 pour-cent, une section transversale des pores de 0,1 à 5 μm et une taille des cellules de pores de 1 à 30 μm , ledit polymère poreux contenant un vinylalcoxysilane; et

55 (iii) un générateur de radicaux libres; et

(b) on chauffe et on mélange (1), (ii) et (iii) conjointement à une température supérieure au point de fusion cristalline de ladite polyoléfine de base.

22. Procédé de production d'un polymère réticulé comprenant les étapes suivantes :

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(a) on mélange

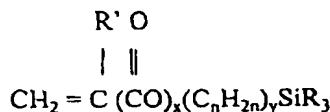
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(i) un polymère de base thermoplastique sélectionné parmi (α) les homopolymères d'éthylène; (β) les copolymères d'éthylène et d'une autre α -oléfine ayant 3 à 6 atomes de carbone; (γ) les copolymères d'éthylène et d'acétate de vinyle, les copolymères d'éthylène et d'acrylate d'éthyle, les copolymères d'éthylène et de 1-butène, de 1-hexène, de 1-octène ou de 4-méthyl-1-pentène;

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(ii) un polymère véhiculaire solide sélectionné parmi les polymères poreux de type éponge, les polymères gonflables et les encapsulats qui ne réagissent pas aisément avec le silane et sont dispersables ou solubles dans le polymère de base, ledit polymère véhiculaire solide étant sensiblement exempt d'eau et contenant un silane ayant pour formule :

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dans laquelle R' représente de l'hydrogène ou un groupe méthyle; x et y sont égaux à 0 ou à 1, pourvu que, lorsque x est égal à 1, y soit égal à 1 également; n est un nombre entier de 1 à 12 inclus; chaque groupe R représente indépendamment un groupe organique hydrolysable sélectionné parmi les groupes alcoxy, aryloxy, aralcoxy, acyloxy aliphatique, oximo, alkylamino et arylamino ou un groupe alkyle inférieur ayant 1 à 6 atomes de carbone inclus, pourvu qu'au maximum un des trois groupes R soit un groupe alkyle; la quantité dudit polymère véhiculaire se situant dans une plage de 0,1 à 20 pour-cent en poids sur la base du poids de l'extrudat et ledit silane étant présent en quantité de 0,1 à 10 pour-cent en poids sur la base du poids total du polymère de base; et

(iii) un générateur de radicaux libres;

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(b) on chauffe et on mélange (1), (ii) et (iii) conjointement à une température supérieure au point de fusion cristalline dudit polymère de base; et

(c) on expose le mélange de (b) à un catalyseur d'hydrolyse/condensation et de l'eau.

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